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## Structure Reports

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***N*-(2-Chloropyrimidin-4-yl)-*N*,2-dimethyl-2*H*-indazol-6-amine**Hao-Fei Qi,<sup>a</sup> Bing-Ni Liu,<sup>b\*</sup> Mo Liu<sup>b</sup> and Deng-Ke Liu<sup>b</sup><sup>a</sup>Materials Science and Engineering, Tianjin Polytechnic University, Tianjin 300160, People's Republic of China, and <sup>b</sup>Tianjin Institute of Pharmaceutical Research, Tianjin 300193, People's Republic of China

Correspondence e-mail: lbn\_1111@yahoo.com.cn

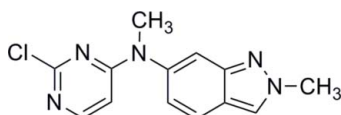
Received 11 October 2010; accepted 21 October 2010

Key indicators: single-crystal X-ray study;  $T = 113$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.036;  $wR$  factor = 0.100; data-to-parameter ratio = 13.3.

In the title compound,  $\text{C}_{13}\text{H}_{12}\text{ClN}_5$ , which is a derivative of the antitumor agent pazopanib {systematic name: 5-[[4-[(2,3-dimethyl-2*H*-indazol-6-yl)methylamino]-2-pyrimidinyl]-amino]-2-methylbenzolsulfonamide}, the indazole and pyrimidine fragments form a dihedral angle of  $62.63(5)^\circ$ . In the crystal, pairs of molecules related by twofold rotational symmetry are linked into dimers through  $\pi$ - $\pi$  interactions between the indazole ring systems [centroid-centroid distance =  $3.720(2)$  Å]. Weak intermolecular C—H...N hydrogen bonds further assemble these dimers into columns propagated in [001].

## Related literature

For background to the pharmacokinetics and clinical studies of the antitumor agent pazopanib, see: Limvorasak & Posadas (2009); Sloan & Scheinfeld 2008; Sonpavde *et al.* (2007). For the synthesis of pazopanib, see: Sorbera *et al.* (2006).



## Experimental

## Crystal data

 $\text{C}_{13}\text{H}_{12}\text{ClN}_5$  $M_r = 273.73$ 

Monoclinic,  $C2/c$   
 $a = 21.432(4)$  Å  
 $b = 9.836(2)$  Å  
 $c = 12.542(3)$  Å  
 $\beta = 90.25(3)^\circ$   
 $V = 2644.1(9)$  Å<sup>3</sup>

$Z = 8$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.28$  mm<sup>-1</sup>  
 $T = 113$  K  
 $0.20 \times 0.18 \times 0.12$  mm

## Data collection

Rigaku Saturn CCD area-detector diffractometer  
 Absorption correction: multi-scan (*CrystalClear*; Rigaku/MS, 2005)  
 $T_{\min} = 0.946$ ,  $T_{\max} = 0.967$

10576 measured reflections  
 2323 independent reflections  
 1982 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.043$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.100$   
 $S = 1.01$   
 2323 reflections

175 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.21$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.25$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C13}-\text{H13B}\cdots\text{N2}^i$	0.98	2.56	3.517 (2)	166

Symmetry code: (i)  $-x, -y + 1, -z + 2$ .

Data collection: *CrystalClear* (Rigaku/MS, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

The authors thank Mr Hai-Bin Song of Nankai University and Mr Shuai Mu of Tianjin University for their helpful suggestions.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV2775).

## References

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**supplementary materials**

*Acta Cryst.* (2010). E66, o2955 [ doi:10.1107/S1600536810042753 ]

## *N*-(2-Chloropyrimidin-4-yl)-*N*,2-dimethyl-2*H*-indazol-6-amine

H.-F. Qi, B.-N. Liu, M. Liu and D.-K. Liu

### Comment

Pazopanib is an oral, second-generation multi-targeted tyrosine kinase inhibitor that targets VEGFR, platelet-derived growth factor receptor and c-kit, key proteins responsible for tumor growth and survival (Limvorasak *et al.*, 2009; Sloan *et al.*, 2008; Sonpavde *et al.*, 2007). The crystal structure of the title compound (I), a derivative of pazopanib, synthesized through the transformation of pazopanib (Sorbera *et al.*, 2006), is reported here.

In (I) (Fig. 1), the indazole and pyrimidine fragments form a dihedral angle of 62.63 (5)°. In the crystal structure, The  $\pi$ - $\pi$  contacts between the indazole systems from the adjacent molecules (Table 1) link them into dimers. Weak intermolecular C—H $\cdots$ N hydrogen bonds (Table 2) link further the dimers into columns propagated in direction [001].

### Experimental

To a stirred solution of the *N*-(2-chloropyrimidin-4-yl)-2-methyl-2*H*-indazol-6-amine 5 g (0.02 mol) in DMF (30 ml) was added Cs<sub>2</sub>CO<sub>3</sub> 9.8 g (0.03 mol) and iodomethane 2.5 ml (5.7 g, 0.04 mol) at room temperature. The mixture was stirred for 5 h. The reaction mixture was then poured into an ice-water bath, and the precipitate was collected *via* filtration and washed with water. The precipitate was air-dried to get off-white solid as crude product. The solid was dissolved in ethyl acetate 30 ml at 278 k, then white crystals were generated slowly.

### Refinement

C-bound H atoms were geometrically positioned (C—H 0.95–0.98 Å), and refined as riding with  $U_{\text{iso}} = 1.2\text{--}1.5 U_{\text{eq}}(\text{C})$ .

### Figures

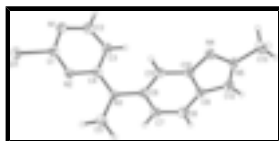


Fig. 1. The molecular structure of (I). Displacement ellipsoids are drawn at the 50% probability level.

## *N*-(2-Chloropyrimidin-4-yl)-*N*,2-dimethyl-2*H*-indazol-6-amine

### Crystal data

C<sub>13</sub>H<sub>12</sub>ClN<sub>5</sub>

$M_r = 273.73$

Monoclinic, *C*2/*c*

$a = 21.432(4)$  Å

$F(000) = 1136$

$D_x = 1.375$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 4286 reflections

# supplementary materials

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$b = 9.836(2) \text{ \AA}$	$\theta = 1.9\text{--}27.9^\circ$
$c = 12.542(3) \text{ \AA}$	$\mu = 0.28 \text{ mm}^{-1}$
$\beta = 90.25(3)^\circ$	$T = 113 \text{ K}$
$V = 2644.1(9) \text{ \AA}^3$	Block, white
$Z = 8$	$0.20 \times 0.18 \times 0.12 \text{ mm}$

## Data collection

Rigaku Saturn CCD area-detector diffractometer	2323 independent reflections
Radiation source: rotating anode confocal	1982 reflections with $I > 2\sigma(I)$
Detector resolution: $7.31 \text{ pixels mm}^{-1}$	$R_{\text{int}} = 0.043$
$\omega$ and $\varphi$ scans	$\theta_{\text{max}} = 25.0^\circ$ , $\theta_{\text{min}} = 1.9^\circ$
Absorption correction: multi-scan ( <i>CrystalClear</i> ; Rigaku/MS, 2005)	$h = -25 \rightarrow 25$
$T_{\text{min}} = 0.946$ , $T_{\text{max}} = 0.967$	$k = -11 \rightarrow 11$
10576 measured reflections	$l = -14 \rightarrow 14$

## Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.036$	H-atom parameters constrained
$wR(F^2) = 0.100$	$w = 1/[\sigma^2(F_o^2) + (0.070P)^2]$
$S = 1.01$	where $P = (F_o^2 + 2F_c^2)/3$
2323 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
175 parameters	$\Delta\rho_{\text{max}} = 0.21 \text{ e \AA}^{-3}$
0 restraints	$\Delta\rho_{\text{min}} = -0.25 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: <i>SHELXL97</i> (Sheldrick, 2008), $F_c^* = kFc[1 + 0.001x Fc^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
	Extinction coefficient: 0.0191 (14)

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	0.227613 (19)	0.06867 (4)	0.79285 (3)	0.0308 (2)
N1	0.13815 (6)	0.18363 (13)	0.68576 (10)	0.0268 (4)
N2	0.18239 (6)	0.30697 (12)	0.83256 (10)	0.0206 (3)
N3	0.14793 (6)	0.52060 (13)	0.88097 (10)	0.0206 (3)
N4	-0.06548 (6)	0.69175 (13)	0.92048 (10)	0.0223 (3)
N5	-0.08472 (6)	0.82333 (13)	0.91397 (10)	0.0218 (3)
C1	0.17633 (7)	0.20335 (15)	0.76669 (12)	0.0211 (4)
C2	0.10043 (8)	0.29266 (17)	0.66821 (13)	0.0284 (4)
H2	0.0720	0.2880	0.6101	0.034*
C3	0.10066 (8)	0.40761 (16)	0.72803 (12)	0.0235 (4)
H3	0.0736	0.4815	0.7124	0.028*
C4	0.14298 (7)	0.41252 (15)	0.81461 (12)	0.0189 (4)
C5	0.19380 (7)	0.52028 (18)	0.96815 (13)	0.0295 (4)
H5A	0.1776	0.4671	1.0280	0.044*
H5B	0.2016	0.6139	0.9915	0.044*
H5C	0.2329	0.4797	0.9432	0.044*
C6	0.10170 (7)	0.62570 (16)	0.88222 (11)	0.0194 (4)
C7	0.12160 (7)	0.76193 (16)	0.86613 (13)	0.0252 (4)
H7	0.1644	0.7801	0.8529	0.030*
C8	0.07992 (7)	0.86753 (17)	0.86940 (13)	0.0267 (4)
H8	0.0935	0.9585	0.8592	0.032*
C9	0.01641 (7)	0.83811 (15)	0.88831 (12)	0.0209 (4)
C10	-0.00297 (7)	0.70070 (15)	0.90426 (11)	0.0188 (4)
C11	0.04067 (7)	0.59365 (15)	0.90219 (12)	0.0195 (4)
H11	0.0282	0.5023	0.9142	0.023*
C12	-0.03903 (7)	0.91238 (17)	0.89500 (12)	0.0246 (4)
H12	-0.0434	1.0080	0.8875	0.030*
C13	-0.15098 (7)	0.85326 (18)	0.92276 (13)	0.0286 (4)
H13A	-0.1573	0.9519	0.9201	0.043*
H13B	-0.1667	0.8179	0.9906	0.043*
H13C	-0.1735	0.8102	0.8636	0.043*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0287 (3)	0.0271 (3)	0.0365 (3)	0.00976 (17)	-0.00266 (19)	-0.00407 (17)
N1	0.0329 (8)	0.0226 (8)	0.0250 (8)	0.0042 (6)	-0.0049 (6)	-0.0020 (6)
N2	0.0177 (7)	0.0222 (8)	0.0220 (7)	0.0014 (6)	0.0025 (5)	0.0001 (6)
N3	0.0176 (7)	0.0228 (7)	0.0214 (7)	0.0028 (6)	-0.0009 (5)	-0.0045 (6)
N4	0.0207 (7)	0.0194 (8)	0.0267 (8)	0.0035 (6)	0.0006 (6)	0.0018 (5)
N5	0.0223 (7)	0.0204 (7)	0.0226 (7)	0.0049 (6)	-0.0011 (5)	0.0004 (6)
C1	0.0205 (8)	0.0200 (9)	0.0228 (9)	0.0009 (7)	0.0050 (7)	0.0024 (7)
C2	0.0344 (9)	0.0284 (10)	0.0225 (9)	0.0018 (8)	-0.0080 (7)	0.0005 (7)
C3	0.0274 (9)	0.0220 (9)	0.0210 (8)	0.0041 (7)	-0.0025 (7)	0.0036 (7)

## supplementary materials

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C4	0.0181 (8)	0.0203 (9)	0.0184 (8)	-0.0010 (6)	0.0052 (6)	0.0028 (6)
C5	0.0229 (9)	0.0349 (10)	0.0307 (10)	0.0057 (8)	-0.0078 (7)	-0.0107 (8)
C6	0.0206 (8)	0.0214 (9)	0.0163 (8)	0.0019 (7)	-0.0011 (6)	-0.0024 (6)
C7	0.0212 (8)	0.0254 (9)	0.0290 (9)	-0.0040 (7)	0.0022 (7)	-0.0015 (7)
C8	0.0278 (9)	0.0194 (9)	0.0328 (10)	-0.0036 (7)	0.0007 (7)	0.0011 (7)
C9	0.0239 (8)	0.0188 (8)	0.0200 (8)	-0.0002 (7)	-0.0013 (6)	0.0001 (6)
C10	0.0201 (8)	0.0199 (8)	0.0164 (8)	-0.0004 (6)	-0.0025 (6)	-0.0002 (6)
C11	0.0225 (8)	0.0176 (8)	0.0185 (8)	-0.0005 (6)	-0.0001 (6)	-0.0003 (6)
C12	0.0303 (10)	0.0180 (8)	0.0255 (9)	0.0009 (7)	-0.0013 (7)	0.0003 (7)
C13	0.0223 (9)	0.0308 (10)	0.0328 (10)	0.0082 (7)	0.0023 (7)	0.0053 (7)

### Geometric parameters (Å, °)

C11—C1	1.7515 (16)	C5—H5B	0.9800
N1—C1	1.315 (2)	C5—H5C	0.9800
N1—C2	1.360 (2)	C6—C11	1.370 (2)
N2—C1	1.3182 (19)	C6—C7	1.421 (2)
N2—C4	1.3564 (19)	C7—C8	1.371 (2)
N3—C4	1.3540 (19)	C7—H7	0.9500
N3—C6	1.4321 (19)	C8—C9	1.413 (2)
N3—C5	1.467 (2)	C8—H8	0.9500
N4—C10	1.3586 (19)	C9—C12	1.398 (2)
N4—N5	1.3607 (17)	C9—C10	1.428 (2)
N5—C12	1.336 (2)	C10—C11	1.409 (2)
N5—C13	1.4551 (19)	C11—H11	0.9500
C2—C3	1.357 (2)	C12—H12	0.9500
C2—H2	0.9500	C13—H13A	0.9800
C3—C4	1.413 (2)	C13—H13B	0.9800
C3—H3	0.9500	C13—H13C	0.9800
C5—H5A	0.9800		
Cg1...Cg2 <sup>i</sup>	3.720 (2)		
C1—N1—C2	112.08 (13)	C11—C6—C7	122.04 (14)
C1—N2—C4	115.36 (12)	C11—C6—N3	119.81 (14)
C4—N3—C6	121.42 (12)	C7—C6—N3	118.11 (13)
C4—N3—C5	120.45 (13)	C8—C7—C6	120.96 (15)
C6—N3—C5	117.04 (12)	C8—C7—H7	119.5
C10—N4—N5	103.19 (12)	C6—C7—H7	119.5
C12—N5—N4	114.34 (13)	C7—C8—C9	118.58 (15)
C12—N5—C13	126.67 (14)	C7—C8—H8	120.7
N4—N5—C13	118.92 (13)	C9—C8—H8	120.7
N1—C1—N2	131.07 (14)	C12—C9—C8	136.31 (15)
N1—C1—C11	114.88 (12)	C12—C9—C10	103.78 (14)
N2—C1—C11	114.05 (11)	C8—C9—C10	119.90 (14)
C3—C2—N1	124.52 (14)	N4—C10—C11	127.55 (14)
C3—C2—H2	117.7	N4—C10—C9	111.71 (13)
N1—C2—H2	117.7	C11—C10—C9	120.74 (14)
C2—C3—C4	117.01 (15)	C6—C11—C10	117.78 (14)
C2—C3—H3	121.5	C6—C11—H11	121.1

C4—C3—H3	121.5	C10—C11—H11	121.1
N3—C4—N2	116.87 (13)	N5—C12—C9	106.98 (14)
N3—C4—C3	123.20 (14)	N5—C12—H12	126.5
N2—C4—C3	119.91 (14)	C9—C12—H12	126.5
N3—C5—H5A	109.5	N5—C13—H13A	109.5
N3—C5—H5B	109.5	N5—C13—H13B	109.5
H5A—C5—H5B	109.5	H13A—C13—H13B	109.5
N3—C5—H5C	109.5	N5—C13—H13C	109.5
H5A—C5—H5C	109.5	H13A—C13—H13C	109.5
H5B—C5—H5C	109.5	H13B—C13—H13C	109.5
C10—N4—N5—C12	0.03 (17)	C11—C6—C7—C8	-0.4 (2)
C10—N4—N5—C13	177.21 (12)	N3—C6—C7—C8	-178.15 (13)
C2—N1—C1—N2	1.9 (2)	C6—C7—C8—C9	-0.5 (2)
C2—N1—C1—C11	-178.80 (12)	C7—C8—C9—C12	-178.16 (16)
C4—N2—C1—N1	-0.3 (2)	C7—C8—C9—C10	0.4 (2)
C4—N2—C1—C11	-179.62 (10)	N5—N4—C10—C11	-179.72 (13)
C1—N1—C2—C3	-1.4 (2)	N5—N4—C10—C9	0.32 (16)
N1—C2—C3—C4	-0.4 (3)	C12—C9—C10—N4	-0.53 (17)
C6—N3—C4—N2	-168.27 (13)	C8—C9—C10—N4	-179.54 (12)
C5—N3—C4—N2	-0.5 (2)	C12—C9—C10—C11	179.50 (13)
C6—N3—C4—C3	13.5 (2)	C8—C9—C10—C11	0.5 (2)
C5—N3—C4—C3	-178.74 (15)	C7—C6—C11—C10	1.3 (2)
C1—N2—C4—N3	179.92 (13)	N3—C6—C11—C10	179.02 (12)
C1—N2—C4—C3	-1.8 (2)	N4—C10—C11—C6	178.69 (14)
C2—C3—C4—N3	-179.73 (15)	C9—C10—C11—C6	-1.3 (2)
C2—C3—C4—N2	2.1 (2)	N4—N5—C12—C9	-0.37 (18)
C4—N3—C6—C11	57.33 (19)	C13—N5—C12—C9	-177.29 (13)
C5—N3—C6—C11	-110.80 (17)	C8—C9—C12—N5	179.28 (16)
C4—N3—C6—C7	-124.86 (16)	C10—C9—C12—N5	0.52 (16)
C5—N3—C6—C7	67.01 (18)		

Symmetry codes: (i)  $-x, y, -z+3/2$ .

*Hydrogen-bond geometry* ( $\text{\AA}, ^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C13—H13B $\cdots$ N2 <sup>ii</sup>	0.98	2.56	3.517 (2)	166

Symmetry codes: (ii)  $-x, -y+1, -z+2$ .

Fig. 1

